## **The nucleophilic 5-***endo-trig* **cyclization of** *gem***-difluoroolefins with homoallylic functional groups: syntheses of ring-fluorinated dihydroheteroaromatics**

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*gem***-Difluoroolefins bearing homoallylic tosylamido, hydroxy, or mercapto groups undergo intramolecular nucleophilic substitution of the nitrogen, oxygen, or sulfur with loss of fluorine** *via* **a 5-***endo***-***trig* **process, leading to 2-fluoro-2-pyrrolines, 5-fluoro-2,3-dihydrofurans, or -thiophenes in high yields.**

5-*endo-trig* cyclization has long been considered to be a geometrically disfavored process in accord with Baldwin's rules.1 Efficient examples of the cyclization, however, have been recently devised in radical-initiated<sup>2</sup> and electrophiledriven3 ring closures. In contrast with these two types of 5-*endo-trig* cyclizations, the corresponding nucleophile-driven ring closure has still rarely been observed in synthetic chemistry.4,5

*gem*-Difluoroolefins possess remarkable reactivity toward nucleophilic substitution for their fluorine atoms *via* addition– elimination processes.6 As one of its applications, we have recently disclosed that β,β-difluorostyrenes bearing *ortho*nitrogen, oxygen, or sulfur heteroatoms readily undergo nucleophilic 5-*endo-trig* cyclization to provide ring-fluorinated heteroaromatics: 2-fluorinated indoles, benzo[*b*]furans, and benzo[*b*]thiophenes.5 Such unique reactivity of *gem*-difluoroolefins is exerted presumably due to (i) the highly polarized C– C double bond (significant single bond character implied by 13C NMR: *ca.* 150 and 90 ppm for  $CF_2=C$ ), which would allow initial 5-membered ring formation, and (ii) the successive elimination of fluoride ion suppressing the reverse ring opening. In this ring-forming reaction the substrates had a benzene ring as an sp2-carbon linker between the nucleophilic functional group and the difluoroolefin part.

In order to broaden the scope of this nucleophilic 5-*endo-trig* cyclization and rule out the possibility of an  $6\pi$ -electrocyclization mechanism, we investigated the reaction of *gem*-difluoroolefins **1**–**3** bearing an *N*-, *O*-, or *S*-functional group which was linked by two sp<sup>3</sup> carbons to the olefinic carbon. Herein we report a facile synthesis of selectively ringfluorinated pyrroline **5**, dihydrofuran **4**, and dihydrothiophene **6** (Scheme 1), which are difficult to access despite the potential uses of ring-fluorinated heterocycles as components of agrochemicals, pharmaceuticals, and dyestuffs.7

*gem*-Difluoroolefin substrates were designed to bear a nucleophilic nitrogen, oxygen, or sulfur atom at the homoallylic position suitable for substitution in a 5-*endo-trig* fashion. To introduce a functional group (HY) at the *gem*-difluoro-



**Scheme 1** Nucleophilic 5-*endo*-*trig* cyclization of *gem*-difluoroolefins **1**–**3**.

homoallylic position, we tried the regioselective hydroboration of 1,1-difluoro-1,3-dienes **7**, which were readily prepared from 2,2,2-trifluoroethyl toluene-*p*-sulfonate and vinyl halides in a one-pot operation according to our method.8 The electron-rich, non-fluorinated double bond in **7** might be more reactive toward borane reagents. Treatment of difluorodienes **7** with 9-borabicyclo[3.3.1]nonane (9-BBN) under reflux in THF selectively promoted hydroboration as expected to generate difluorohomoallylboranes, and successive treatment with alkaline aqueous hydrogen peroxide gave *gem*-difluorohomoallyl alcohols **1a**–**c** in good yields (Scheme 2). The corresponding nitrogen- and sulfur-containing substrates **2a**–**e**, **3a**,**b** were easily obtained from **1** as shown in Scheme 2.

The cyclization of homoallyl alcohols **1a**,**b** was attempted by treatment with 1.2 eq. of NaH in several solvents. While the use DMPU or NMP gave no cyclized products, DMF, DMA, or DMSO successfully promoted the 5-*endo-trig* cyclization to afford 5-fluoro-2,3-dihydrofurans **4a**,**b** in good yields (Table 1, Entries 1,2).9† KH was less effective than NaH, and highdilution conditions ( $[1a] = 0.03$  mol L<sup>-1</sup>) raised the yield by 10% compared to the case of  $[\text{1a}] = 0.2 \text{ mol } L^{-1}$ .

Moreover, we examined the 5-*endo-trig* cyclization of the substrates with *N*-nucleophiles under similar conditions. Whereas *N*-unsubstituted and *N*-butylhomoallylamines **2c**,**d** did not cyclize, the *N*-phenyl substrate **2e** afforded 4-methyl-1-phenyl-3-(3-phenylpropyl)-2-pyrrolidone **8** *via* hydrolysis of



*tions*: i, 9-BBN (1.1-1.4 eq.), THF, reflux, 6-7 h; ii, aq.H<sub>2</sub>O<sub>2</sub>, aq. NaOH, 0 °C, 2 h; iii, BocNHTs (1 eq.), PPh<sub>3</sub> (1 eq.), EtO<sub>2</sub>CN=NCO<sub>2</sub>Et (1 eq.), THF, rt, 2 h; iv,  $CF_3CO_2H$  (15 eq.),  $CH_2Cl_2$ , rt, 2 h; v, phthalimide (1 eq.),  $PPh_3$  (1 eq.), EtO<sub>2</sub>CN=NCO<sub>2</sub>Et (1 eq.), THF, rt, 2 h; vi, NH<sub>2</sub>NH<sub>2</sub> $\cdot$ H<sub>2</sub>O (2 eq.), EtOH, reflux, 2 h; vii, TsCl (1 eq.), Py, rt, 8 h; viii, Bu<sup>n</sup>NH<sub>2</sub> (26 eq.), reflux, 6 h; ix, PhNH<sub>2</sub> (26 eq.), reflux, 6 h; x, AcSNa (1 eq.), DMF, 70 °C, 3 h; xi,  $K_2CO_3$  (1 eq.), MeOH, 0 °C, 1 h.

**Table 1** Synthesis of ring-fluorinated dihydroheteroaromatics **4**–**6***a*

	Entry	YH	R <sup>1</sup>	$R^2$	Substrate	Base (eq.)	Time	Product	Yield $(\%)^b$
		ΟH	CH <sub>2</sub> CH(Me)Ph	Me	$1a^c$	NaH $(1.2)$	7 h	$4a^c$	67
	◠	<b>OH</b>	Bu <sup>n</sup>	CH <sub>2</sub> Ph	1b	NaH (1.2)	11 h	4h	62
		PhNH	(CH <sub>2</sub> ) <sub>3</sub> Ph	Me	2e	NaH (1.1)	24 h	8	62
	4	<b>TsNH</b>	CH <sub>2</sub> CH(Me)Ph	Me	$2b^c$	NaH $(1.1)$	4 d	5 <sup>c</sup>	80
	5	SH	$(CH_2)_3$ Ph	Me	3 <sub>b</sub>	NaH(1.1)	4 h	6	76
	6	SC(O)Me	$(CH_2)_3$ Ph	Me	3a	NaOMe $(1.0)$	8 h	h	69
<sup><i>a</i></sup> The reaction was conducted in DMF at 90 °C. [Substrate] = 0.02–0.03 mol L <sup>-1</sup> , b Isolated yield, c 1:1 Diastereomer mixture.									



**Scheme 3** Competitive cyclization: 5-*endo-trig vs*. 5-*exo-trig Reagents and conditions*: i, NaH (1.1 eq.), DMF, 100 °C, 0.3 h.

the expected 2-fluoropyrroline (Entry 3). The *N*-4,4-difluorobut-3-enyltoluene-*p*-sulfonamide **2b** underwent the desired ring closure to give 2-fluorinated pyrroline **5** in 80% yield (Entry 4).10‡

As a further example of the cyclization, the intramolecular substitution of sulfur nucleophiles (which Baldwin's rules allow for the normally disfavored 5-*endo-trig* process) was also examined.1*b* The reaction of **3b** under similar conditions to those for **1** and **2** provided 5-fluorinated dihydrothiophene **6** in 76% yield (Entry 5).11 In addition, treatment of thioacetic *S*-acid ester **3a** with sodium methoxide allowed the cyclization *via in situ* generated thiolate to give **6** (Entry 6).§

In order to demonstrate the favored nature of 5-*endo-trig* cyclization in *gem*-difluoroolefins, we tried the competitive reaction between 5-*endo-trig* and 5-*exo-trig* processes. The  $\beta$ ,  $\beta$ difluoro- $\alpha$ , $\beta$ -unsaturated ester **9** bearing a 2-toluene-*p*-sulfonamidoethyl group was designed as a substrate which could undergo the Michael reaction and the transacylation *via* 5-*endotrig* and 5-*exo-trig* processes, respectively. Compound **9** was prepared by the  $S_N^2$  reaction of methyl 2-(trifluoromethyl)propenoate12 with benzyloxymethylcopper13 and successive deprotection of the benzyl group, followed by the introduction of an NHTs group according to the same procedure for **2b**. On treatment of **9** with NaH in DMF, the 5-*endo-trig* cyclization proceeded to lead exclusively to the 2-fluorinated pyrroline derivative **10** as shown in Scheme 3.<del></del>

In conclusion, normally 'disfavored' 5-*endo-trig* ring closures are successfully achieved in the intramolecular addition– elimination reaction of *gem*-difluoroolefins bearing nucleophilic heteroatoms linked by two sp3 carbons to the vinylic carbon as well as two sp2 carbons.5 Thus, *gem*-difluorohomoallylamine, alcohol, and thiol derivatives open a new way to the syntheses of selectively ring-fluorinated heterocyclic compounds.

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## **Notes and references**

 $\dagger$  **4b**:  $\delta_F$  (471 MHz, CDCl<sub>3</sub>–C<sub>6</sub>F<sub>6</sub>) 42.4 (1F, s).

- $\ddagger$  **5**:  $\delta_F$  (471 MHz, CDCl<sub>3</sub>–C<sub>6</sub>F<sub>6</sub>) 36.2 (0.5F, s), 36.6 (0.5F, s).
- § **6**:  $\delta_F$  (471 MHz, CDCl<sub>3</sub>-C<sub>6</sub>F<sub>6</sub>) 35.7 (1F, br s).

¶ The fluorine-free analogue, dimethyl 4-methyleneglutamate was reported to cyclize to 5-methoxycarbonyl-3-methylene-2-pyrrolidone *via* 5-*exo-trig* pathway without formation of the 5-*endo-trig* product.1*b*,14  $\parallel$  **10**:  $\delta_F$  (471 MHz, CDCl<sub>3</sub>–C<sub>6</sub>F<sub>6</sub>) 61.8 (1F, t,  $J_{FH}$  = 4.8 Hz).

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